

GDOES depth profile analysis for the investigation of the thermal stability and the degradation mechanisms of Ta based diffusion barriers for Cu metallization systems

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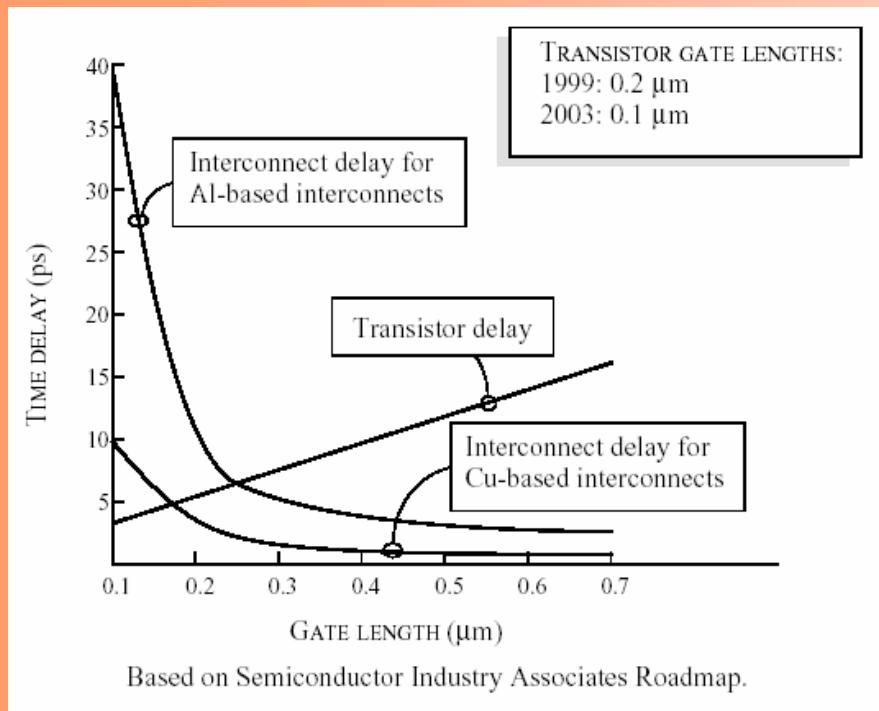
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³ Dresden University of Technology

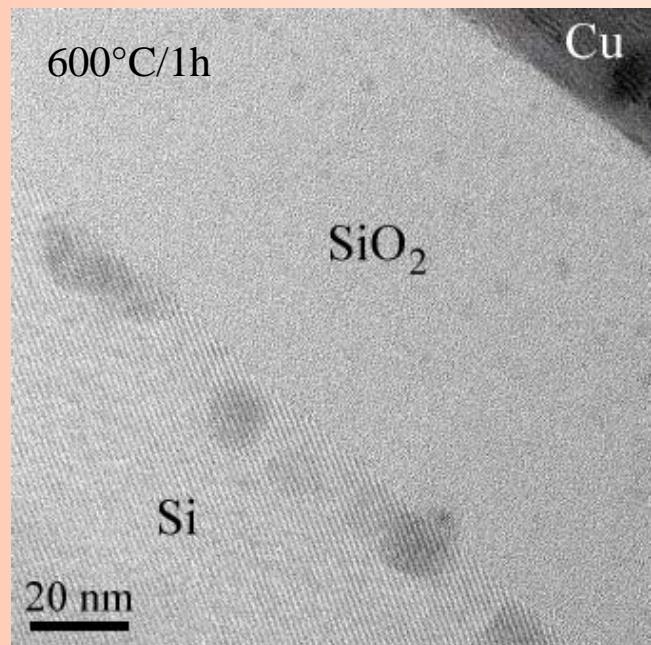
Outline

1. Motivation of the investigations
2. Sample preparation and investigation methods
3. Thermal behavior of various metallization systems
during annealing:
 - Cu/Ta/SiO₂/Si
 - Cu/TaN/SiO₂/Si
 - Cu/Ta/TaN/Ta/SiO₂/Si
 - Cu/Ta₇₃Si₂₇/SiO₂/Si
4. Summary

Advantages and disadvantages of copper as interconnect material



Cu/SiO₂/Si - TEM

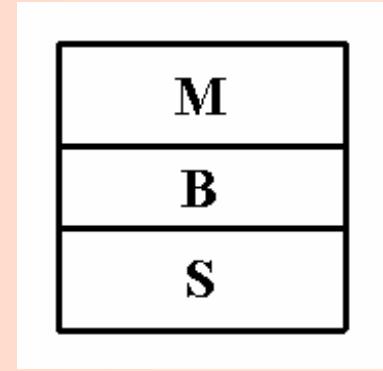


Compared to aluminum:

- ☺ Lower electrical resistivity
- ☺ Higher thermal conductivity
- ☺ Higher electromigration resistance
- ☺ Better stress migration behavior
- ☹ High diffusivity of Cu in Si and SiO₂
- ☹ Cu₃Si formation at $T \approx 200^\circ\text{C}$
- ⇒ Need for diffusion barriers between Cu and interlayer dielectric

Barrier layer requirements

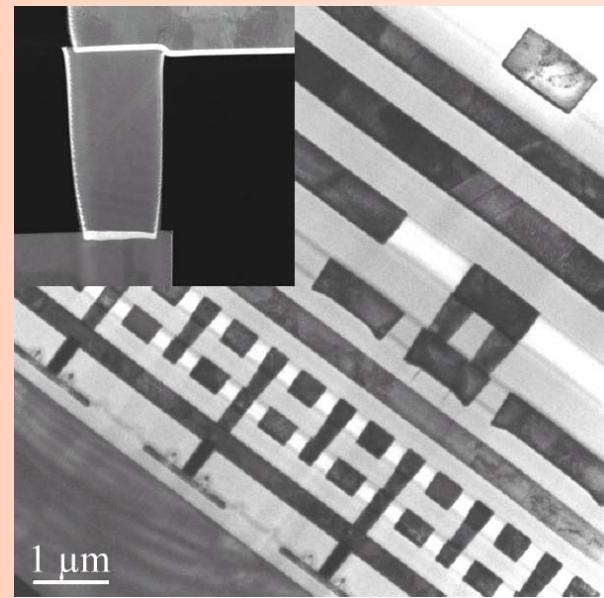
- Defect-free microstructure of B and its thermal stability
- Low diffusion coefficients of M in B and of S in B
- High chemical stability of the interfaces between M and B as well as between S and B
- High electrical conductivity of B
- High thermal conductivity of B
- Good adhesion of B on S as well as of M on B
- Good mechanical properties (low residual stresses)
- Conformal deposition of B



Tantalum diffusion barriers:

- ☺ High melting point ($T_{m,Ta} = 3020 \text{ }^{\circ}\text{C}$)
- ☺ High chemical stability of Ta/Si and Ta/SiO₂ interfaces
- ☺ Very low solubility of Cu in Ta and vice versa
- ☺ No reaction between Cu and Ta
- ☺ High electrical conductivity (especially for α -Ta phase)
- ☺ Polycrystalline microstructure \Rightarrow Cu diffusion

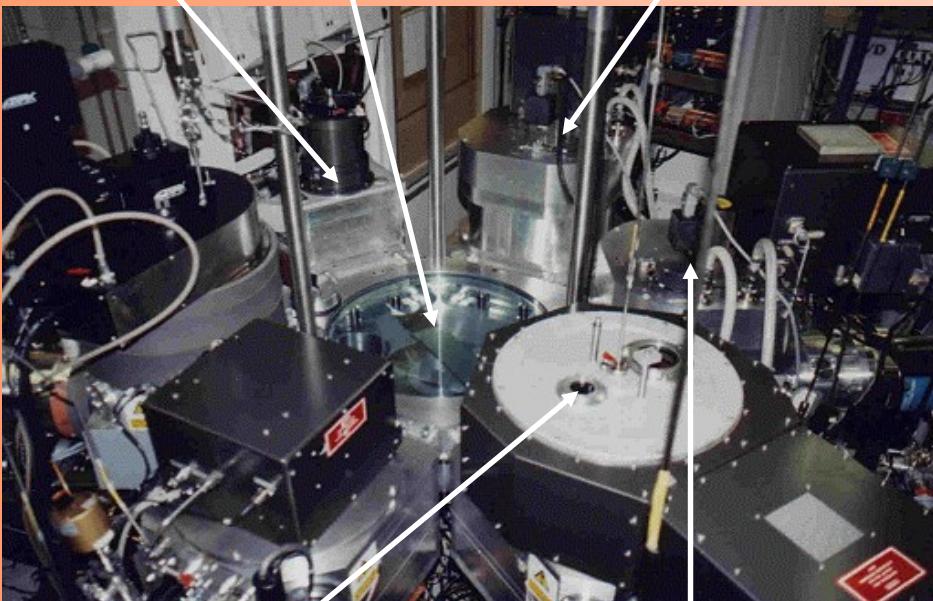
⇒ Need for more stable diffusion barriers



Sample preparation

Cluster Tool

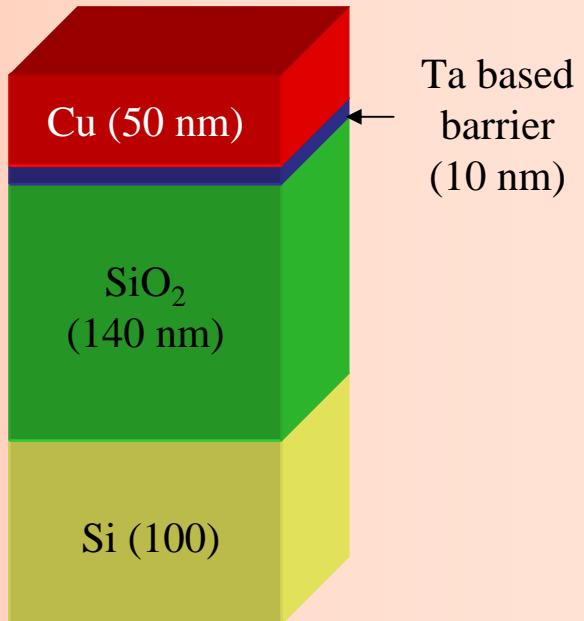
Load lock Dealer



ICP soft etch chamber
Ar plasma, $P = 200$ W

PVD copper chamber

$p_0 < 2 \times 10^{-5}$ Pa, d.c. magnetron sputtering, $P = 1000$ W



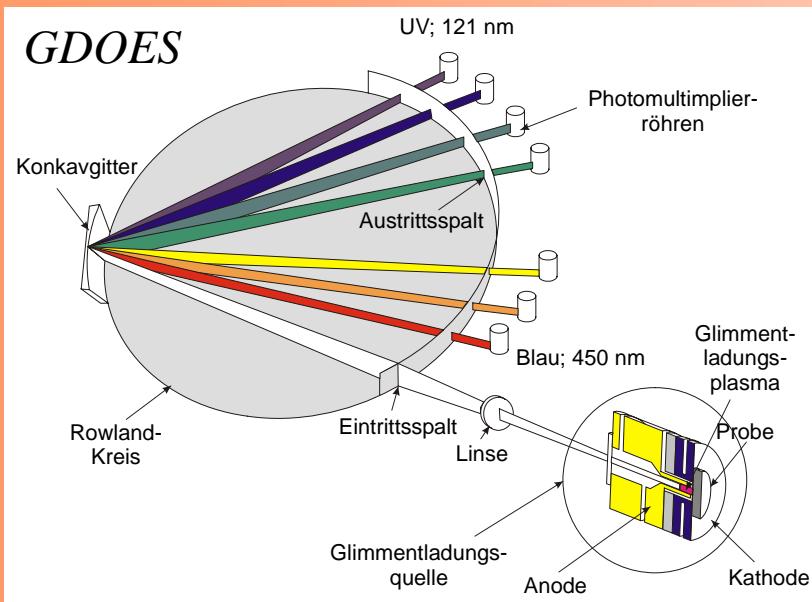
PVD barrier chamber

$p_0 < 2 \times 10^{-5}$ Pa, Ta / Ta₅Si₃ target, r.f. magnetron sputtering, $P = 1000$ W, different N₂/Ar flow ratios

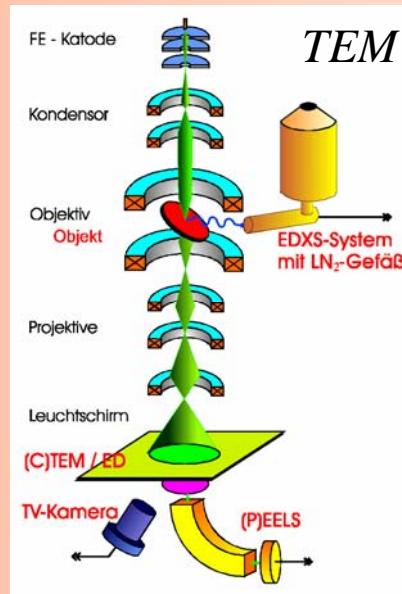
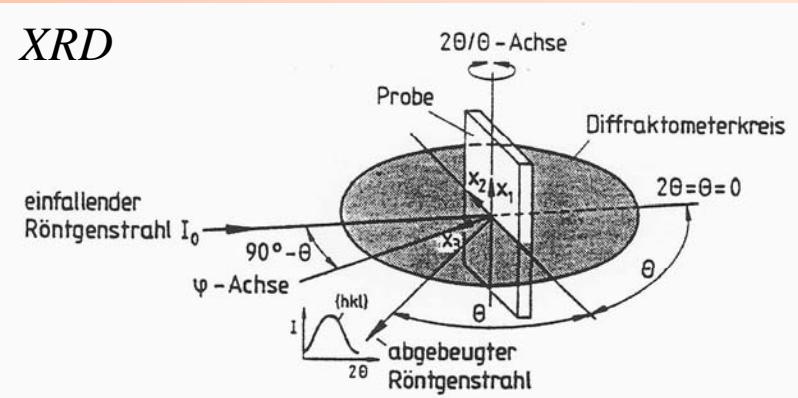
After sputter deposition annealing under vacuum conditions ($p \approx 10^{-4}$ Pa) at various temperatures and annealing times

Investigation methods

Glow Discharge Optical Emission Spectroscopy (GDOES) Depth profile analysis

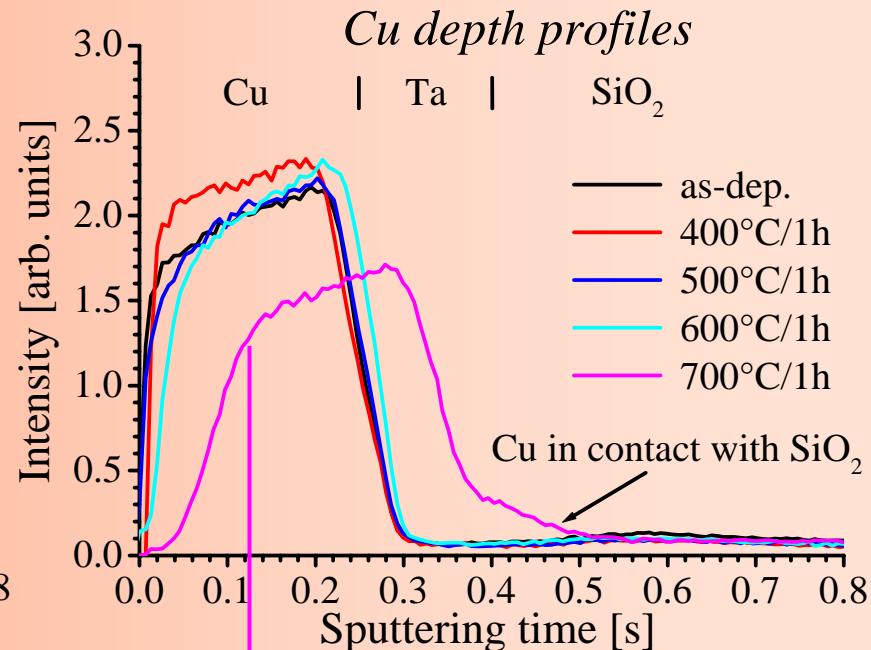
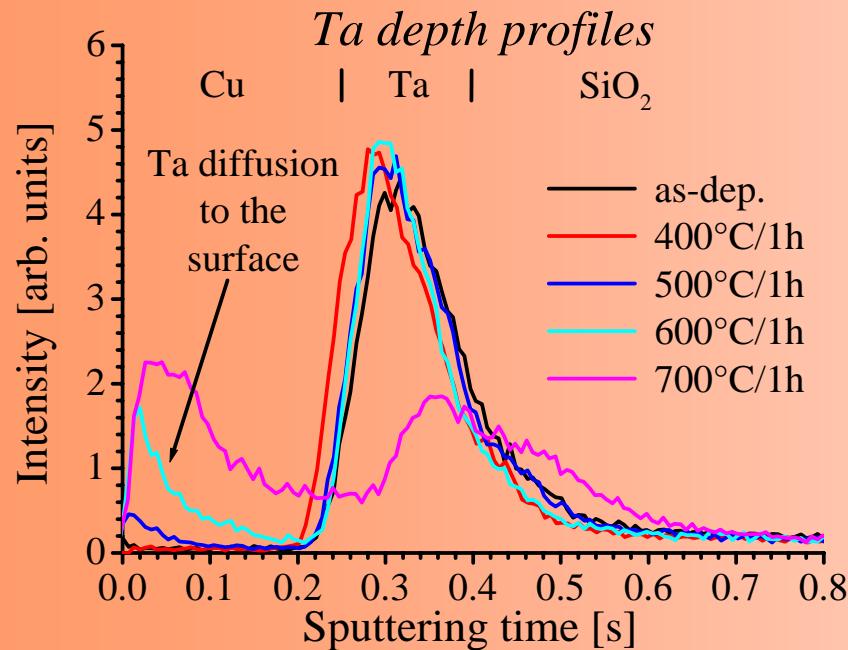


X-Ray Diffraction (XRD) Phase formation behavior

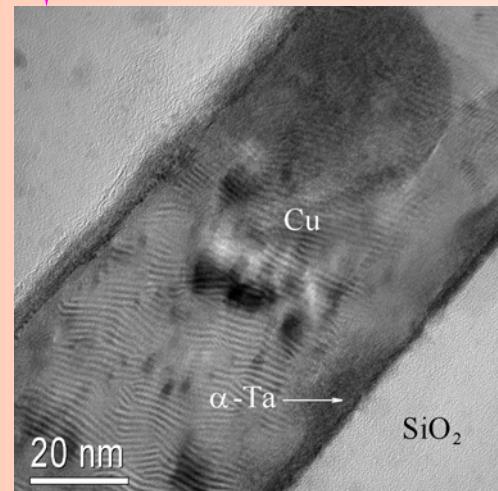


(Analytical) Transmission Electron Microscopy (TEM) Local microstructure

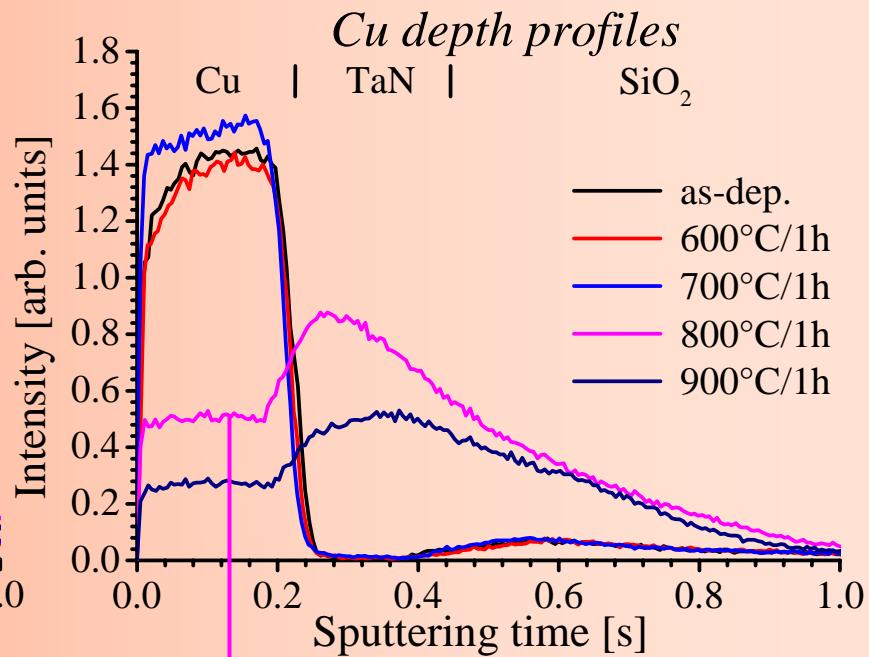
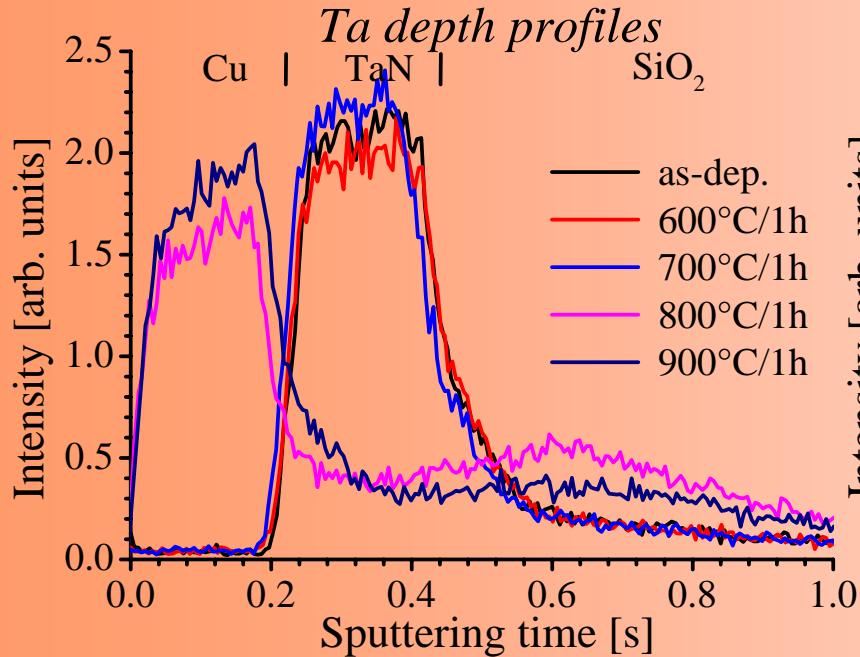
Cu/Ta/SiO₂/Si



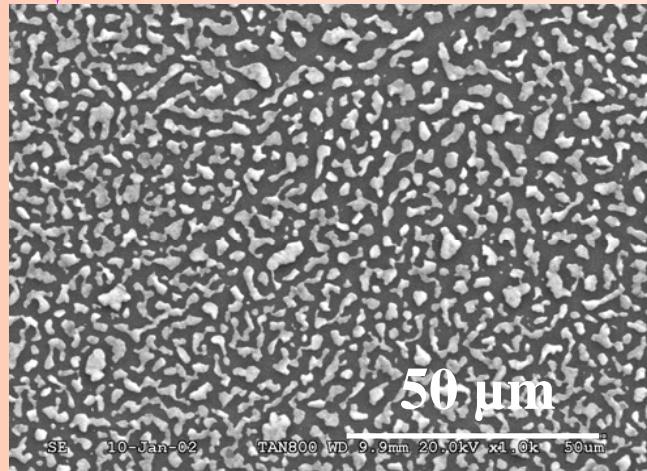
- 500°C/1h: Ta diffusion to the surface
- 700°C/1h: α-Ta crystallite growth,
Cu in contact with SiO₂ ⇒ Cu diffusion



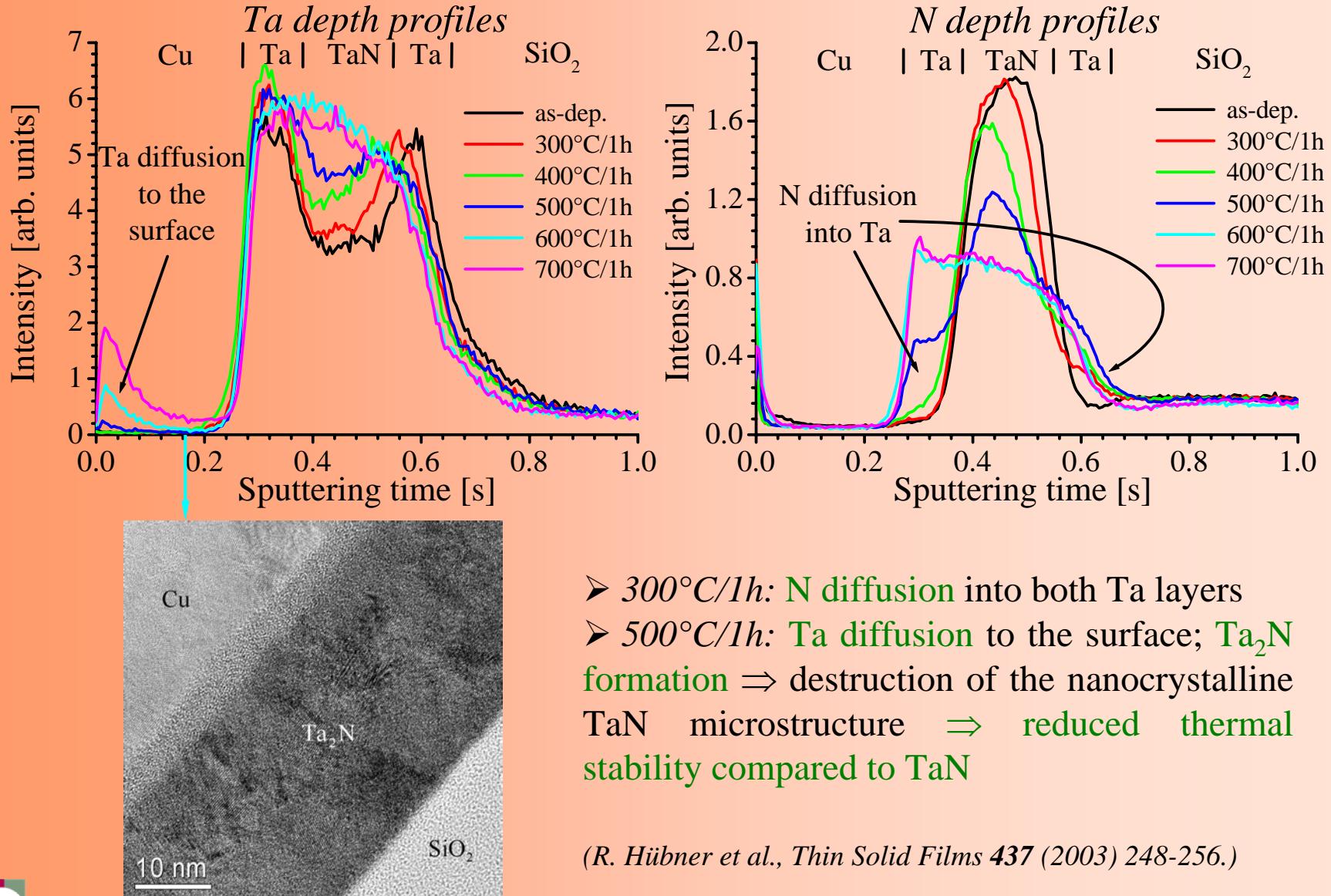
Cu/TaN/SiO₂/Si



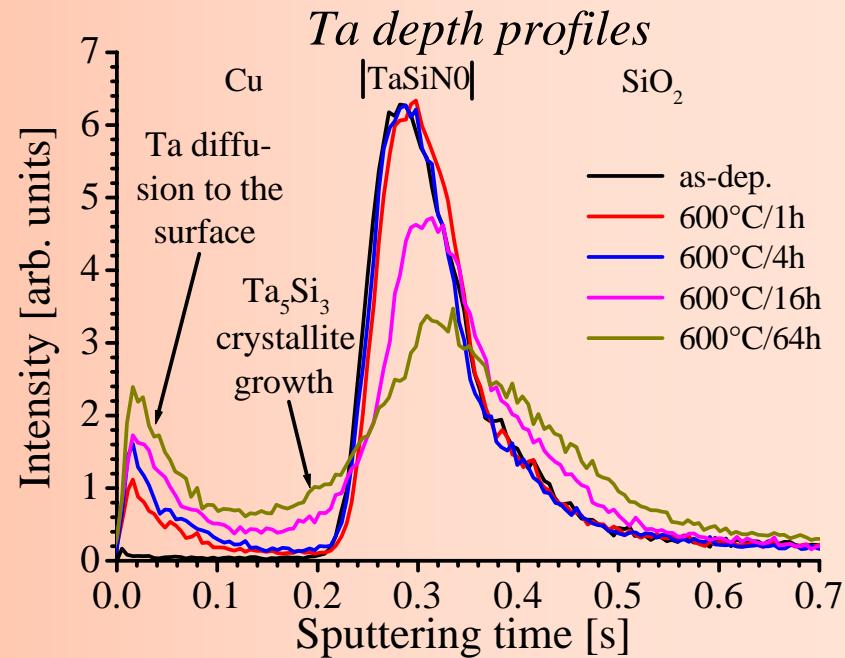
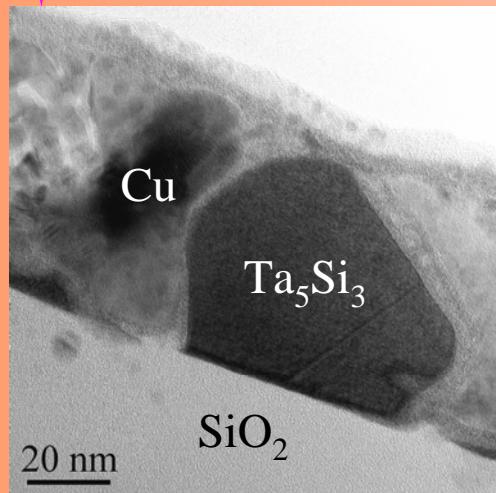
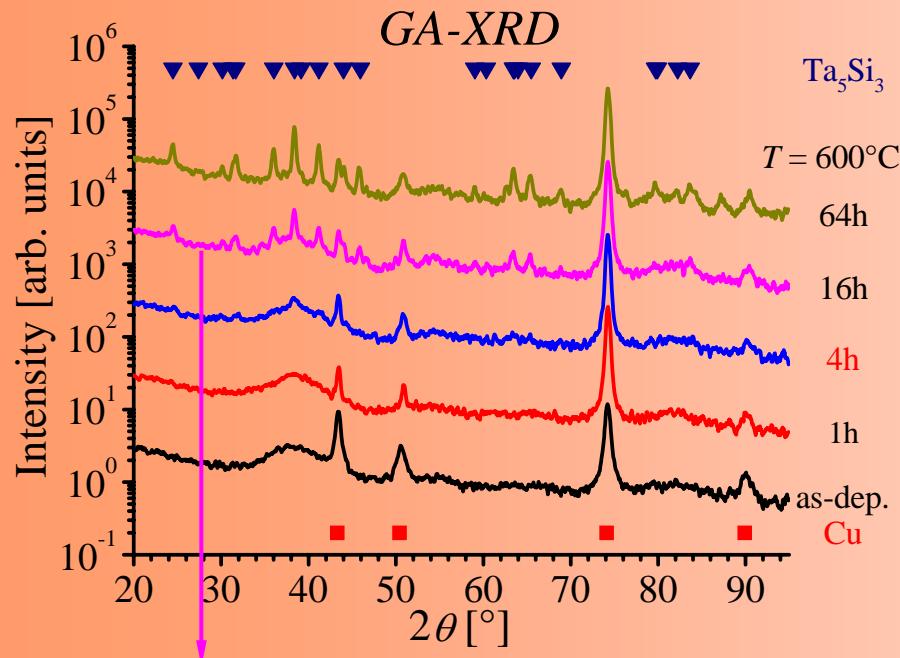
- No Ta diffusion to the surface
- 800°C/1h: **Cu agglomeration** – not to be mistaken for Cu diffusion!



Cu/Ta/TaN/Ta/SiO₂/Si



Cu/Ta₇₃Si₂₇/SiO₂/Si



- 600°C/1h: Ta diffusion to the Cu surface
- 600°C/≥4h: Premature barrier crystallization into Ta₅Si₃; Cu diffusion into SiO₂

(R. Hübner et al., *Thin Solid Films* **458** (2004) 237-245.)

Summary

- GDOES depth profile analysis – simultaneous and quick acquisition of element distributions with sufficiently high depth resolution \Rightarrow quick and comparative characterization of the setup of Cu metallization systems in the as-deposited state and after annealing
- In particular analysis of various diffusion processes: N redistribution within graded Ta-Ta_N barriers, Ta diffusion out of the barrier to the sample surface, Cu diffusion through the barrier into the substrate \Rightarrow in conjunction with other methods (XRD, TEM) determination of the thermal stability and the degradation mechanisms of Ta-based diffusion barriers
- Interpretation of the element depth profiles – separation of methodically caused effects from sample properties