
***Fortschritte bei der Analyse leichter Elemente
mit der Glimmentladungs-Spektrometrie***

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Colloquium Analytische Atomspektroskopie, 23. – 26. September 2019

18. GDS-Anwendertreffen, 26. – 27. September 2019



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Outline

Introduction

(Importance of Light Element Analysis (H, O, N))

Depth resolved, metrological characterization of pure elements (EMRP)

Effects of Light Elements in GDS

Quantification

Preparation and Validation of Sintered Calibration Samples

Calibration and Application

Oxygen

(Hydrogen and Nitrogen at IGDSS 2018 in Berlin)

Summary

Effects of Light Elements in Glow Discharge Spectrometry

- **Plasma physics**, e.g. gases are removed by flow, discharge parameters and excitation/ionization processes may change
Zdenek Weiss, Spectrochim. Acta B61 (2006) 121-133
- **Plasma chemistry**, e.g. dissociation + recombination, combination/gettering.
These processes (e.g. equilibrium of X and X_2) depend on the matrix.
- **Contamination** of gas, source and/or sample, (H_2O , C_xH_y) or by leakages (N_2 , O_2 , CO_2), chemisorption, adsorption, desorption

But many, many applications

- ⇒ 1. Instrumental developments (sources, spectrometers, vacuum system, software tools)
- ⇒ 2. Fundamental investigation (mostly using gas mixtures)

Compare the effect with sputtering of samples containing the light elements!

Absolute Calibration of Light Elements (H, N and O)

$$I_i = f_i (c_i \cdot q) + \text{background} + \text{interference}$$

GD-OES: f_i = Emission Yield factor - constant, if no self-absorption or other effects

GD-MS: f_i =Sensitivity factor

N. Jakubowski et al., J. Anal. At. Spectrom., 1992, 7, 951

R. Muñiz et al., Spectrochim. Acta Part B, 2017, 135, 34

Accurate sputtering rate measurements are essential for calibration.

3 D crater volume: $\Delta V/V \approx 1\%$ possible, usually between 5-10%.

Compact conductive H-, N- and O-calibration samples with different matrix are missing, but essential to check matrix independence (efficient method). Layered material useful, but determination of composition, thickness, density is more difficult and samples are destroyed during calibration.

⇒ Preparation of sintered calibration samples including light elements (H, O, N)

Preparation of Sintered Powder Mixtures

Matrices: Cu, Al, Mg

Compounds: TiH₂, ZrH₂, MgO, Al₂O₃, CuO, Cu₂O, AlN, Si₃N₄ for light elements H, O, N
NaCl, KCl, KBr, RbCl, CaF₂, CaO, LiCl, SrCO₃ for other purpose

*Powder is
Mixed (5 min)*



Sintering conditions

Temperature: 400 °C

Pressure: 150 kN

Inert gas: Ar

*Sintering
Process
(30 min)*



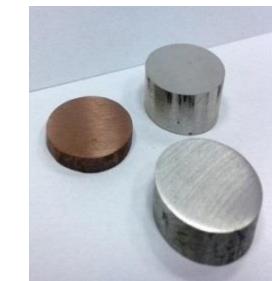
Sintering conditions

Temperature: 600 °C

Pressure: 150 kN

Inert gas: Ar

Sintered
Samples

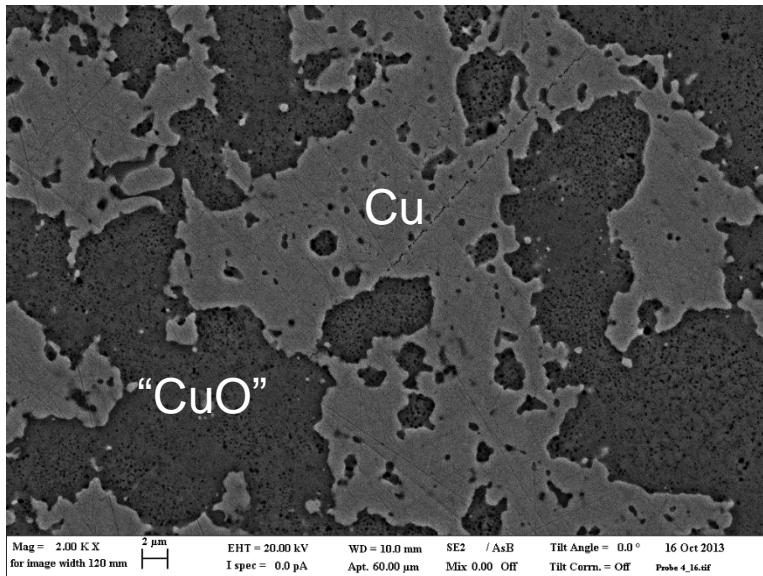


Properties:

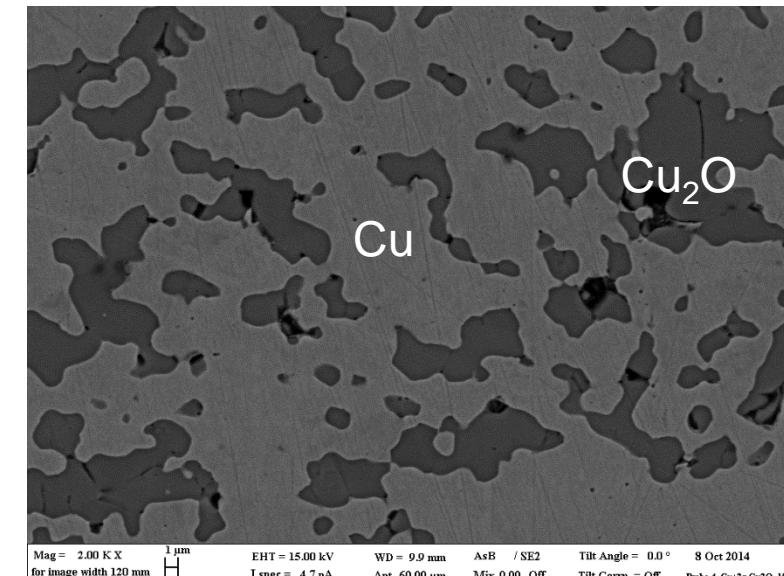
Ø = 20 mm, h ≈ 20 mm, grindable
density > 95% - vacuum tight
homogeneous in mm range - 5% reproducibility

SEM Pictures of Sintered Samples

10 g Cu+2 g CuO ($\triangleq 3.35$ m% O)



10 g Cu+2 g Cu₂O ($\triangleq 1.87$ m% O)

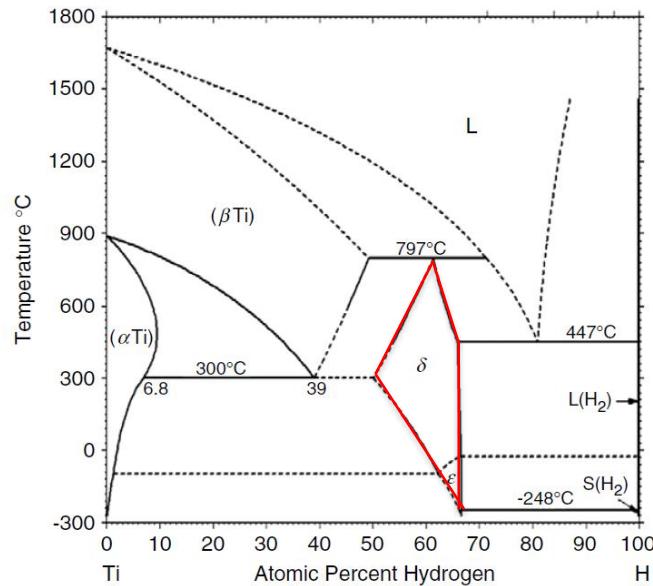


CuO converts into Cu₂O at sintering (XRD)
⇒ low density ($\approx 89\%$) and closed porosity
CGHE: c(O) = 3.58 m%

Cu₂O is stable (XRD), > 99% density
CGHE: c(O) = 1.88 m%
⇒ Validation of concentration is essential

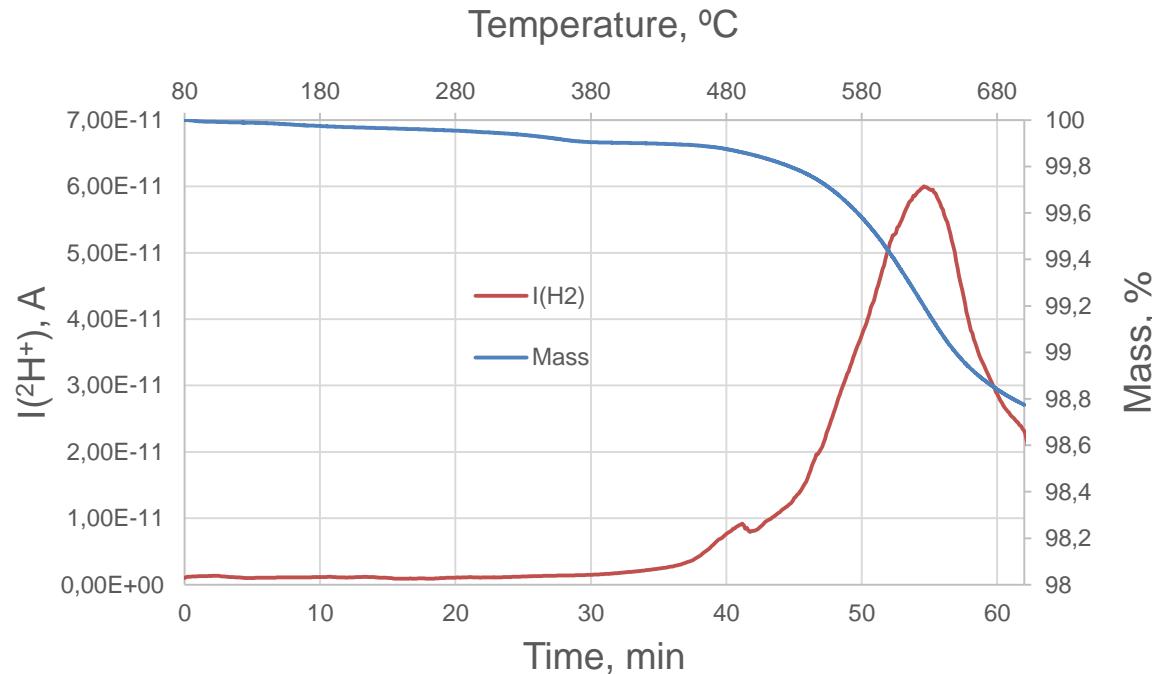
Thermogravimetry (TGA) and Evolved Gas Analysis (EGA)

10 g Cu + 6 g TiH₂, c(H) = 1.5 m%



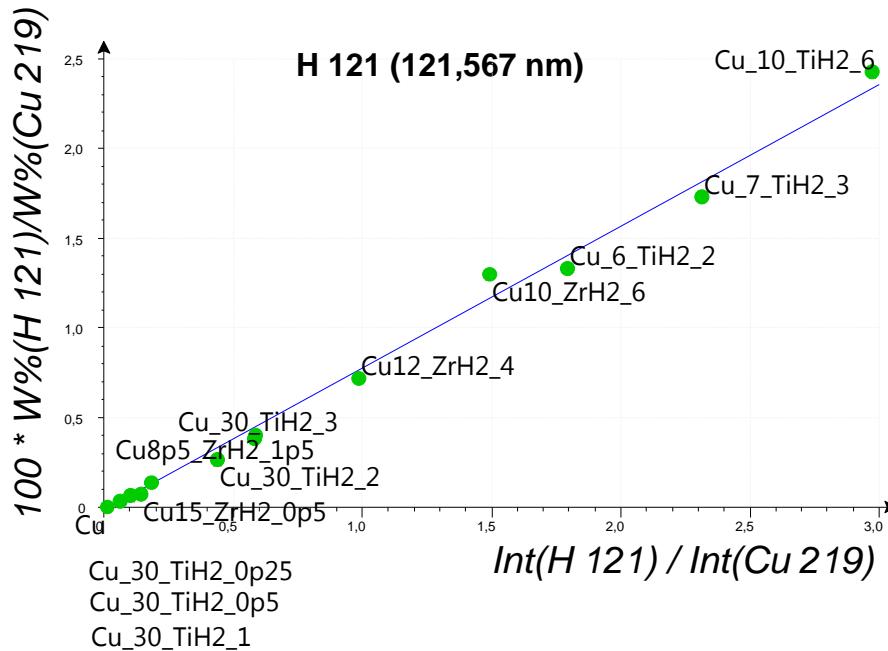
$$T_S(\delta, \text{TiH}_2) = 797 \text{ } ^\circ\text{C}$$

*Yuh Fukai,
J. Japan Inst. Met. 55, 1991, 17*



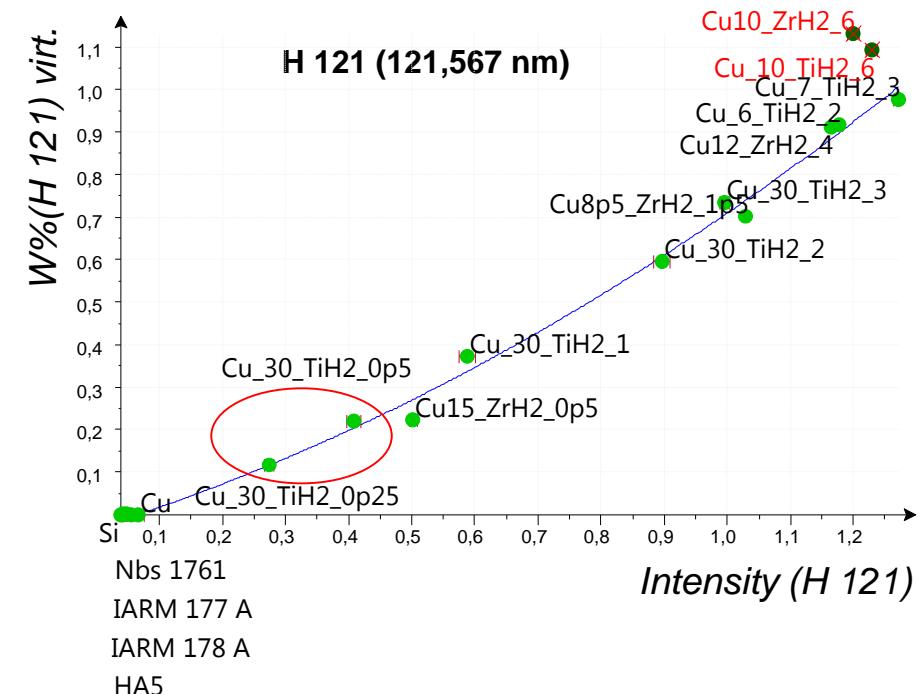
- Decomposition starts at $\approx 450 \text{ } ^\circ\text{C}$.
 - All H is lost at $700 \text{ } ^\circ\text{C}$ and atmospheric pressure!
- ⇒ Validation of concentration is essential

H calibration using sintered Cu+TiH₂ and Cu+ZrH₂ (700 V, 20 mA)



Ratio to Cu 219 compensates saturation and self reversal.

But, quantification is matrix dependent!



Absolute model works up to 1 m% H in Cu(Ti,Zr)

New: $30 \text{ g Cu} + 0.25 \text{ g TiH}_2 \rightleftharpoons c(\text{H}) = 331 \mu\text{g/g}$

$30 \text{ g Cu} + 0.50 \text{ g TiH}_2 \rightleftharpoons c(\text{H}) = 657 \mu\text{g/g}$

For applications with other matrix, e.g. DLC.

Validation by Carrier Gas Hot Extraction

Successfully for TiH_2 , ZrH_2 , CuO , Cu_2O , Si_3N_4 in Cu

Failed at MgO, Al_2O_3 , AlN in Mg and Al

- Al_2O_3 , AlN and MgO with high melting point > 2000°C
- Mg and Al getter efficiently.

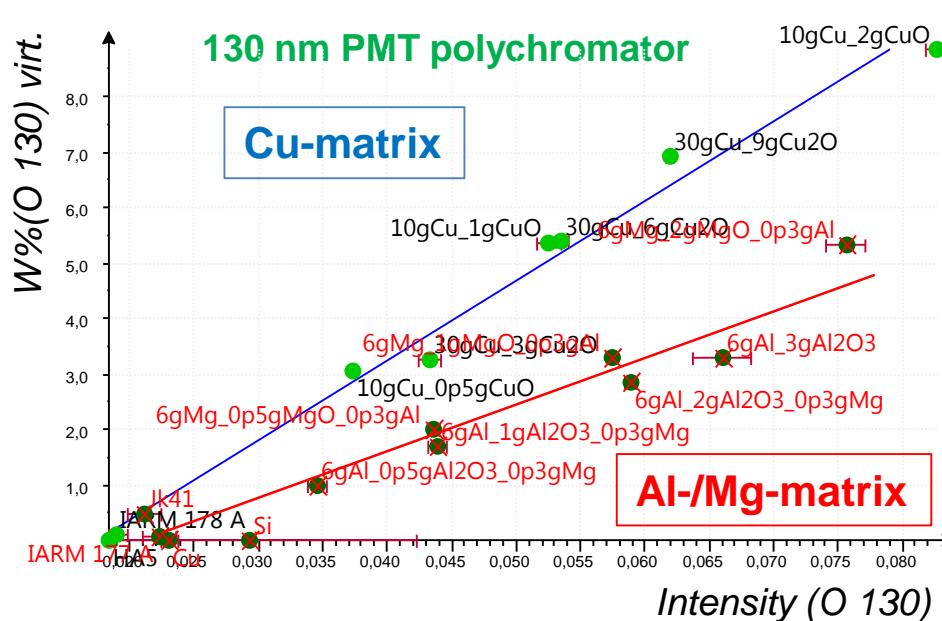
Because

- Decomposition of AlN, Al_2O_3 and MgO is unlikely at 600 °C.
 - Introduction of important amounts of additional H, O and N is not possible.
 - Densities of sintered samples agree with phase densities better than 95%.
- ⇒ Calculated concentrations were used.

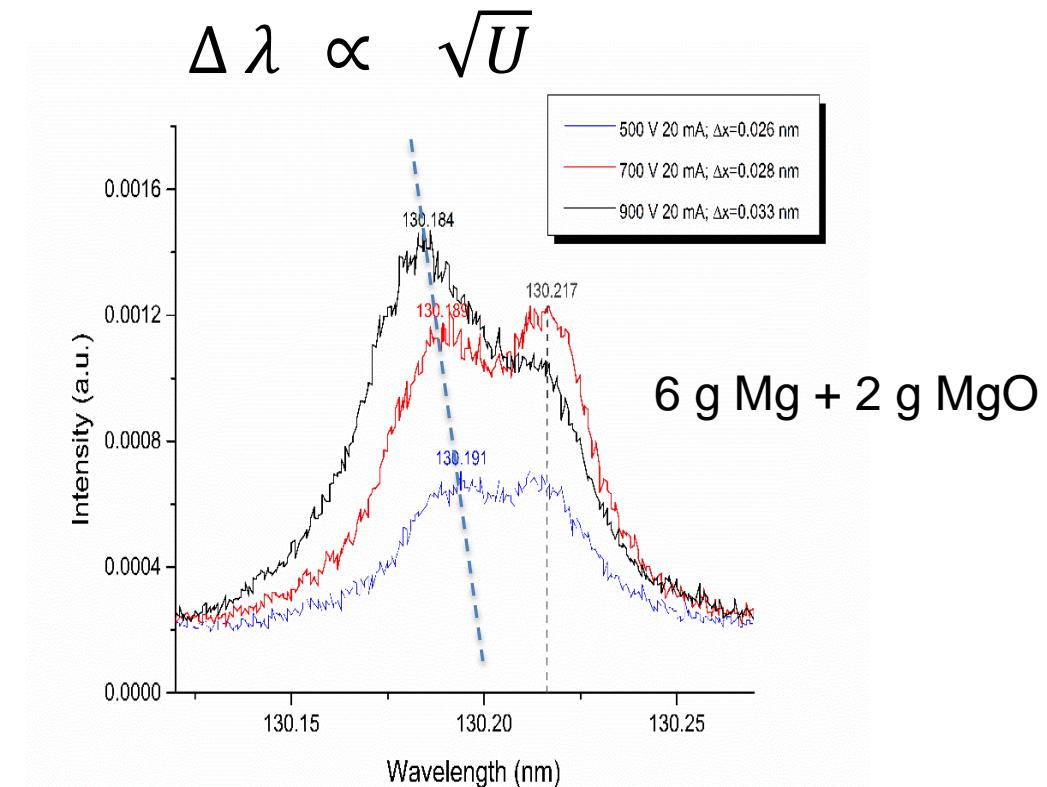
GDS is very welcome for light element analysis, if matrix independent and accurate.

Quantification of Oxygen (700 V, 20 mA)

Line Shift at 130 nm, already reported 2009 at CSI by Michael Köster (Payling Award)



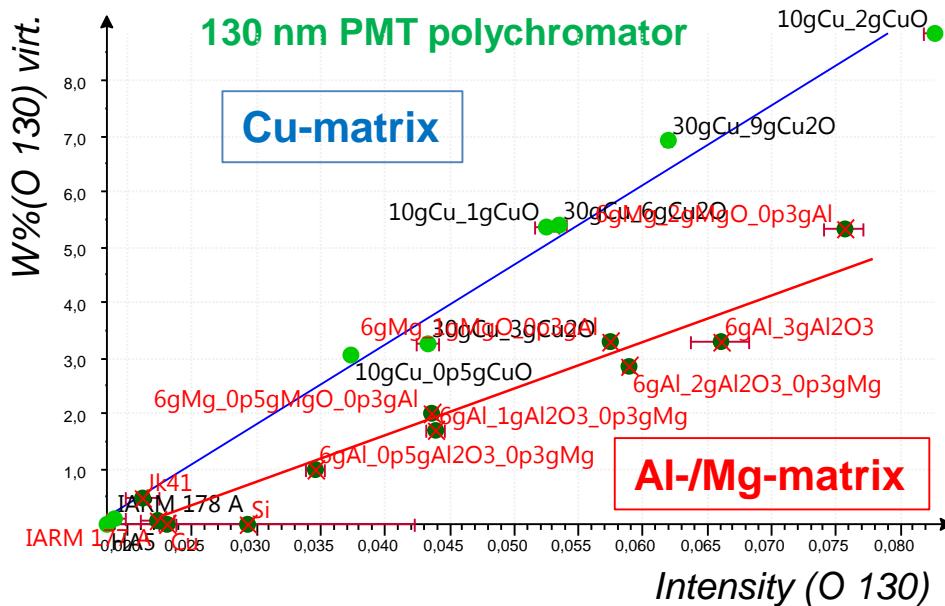
⇒ Matrix dependent for Cu-, Mg- and Al-matrix
(EMRP project SIB09 Elements 2013/14)



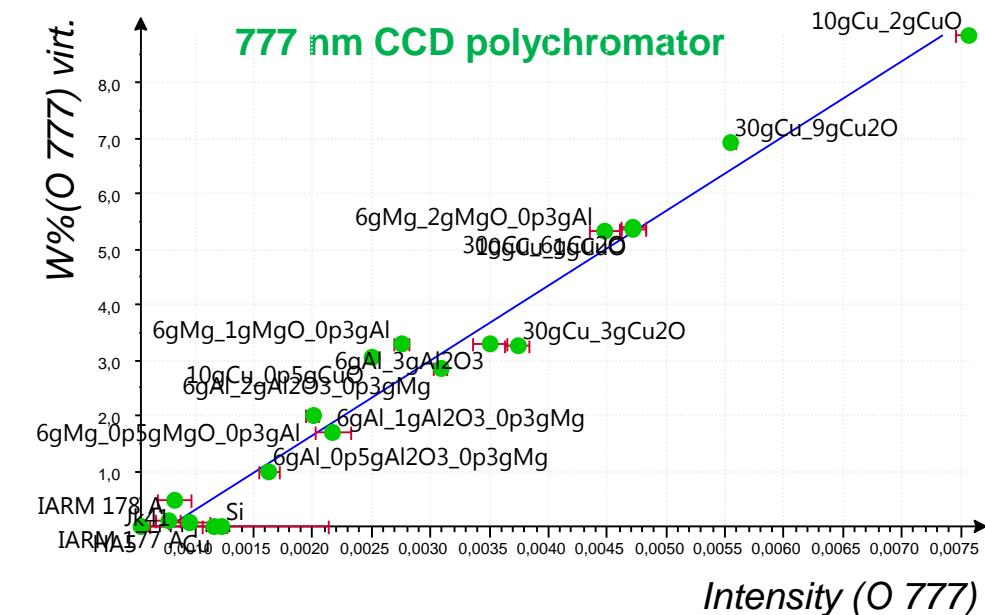
Cristina Gonzalez-Gago et al., The use of matrix-specific calibrations for oxygen in analytical glow discharge spectrometry, Anal Bioanal Chem (2014) 406:7473

Quantification of Oxygen

Line Shift at 130 nm, already reported 2009 at CSI by Michael Köster (Payling Award)

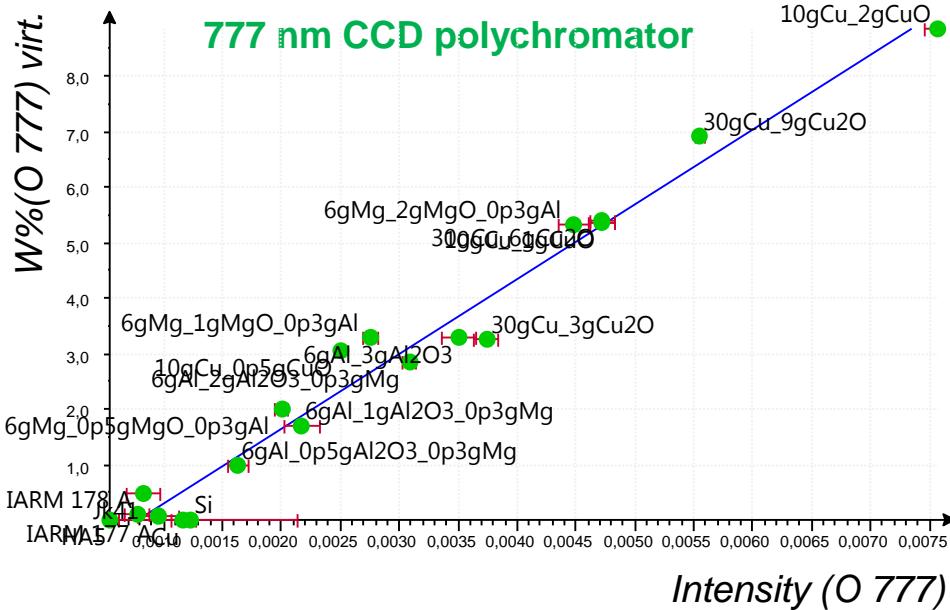


⇒ Matrix dependent for Cu-, Mg- and Al-matrix
(EMRP project SIB09 Elements 2013/14)



Use of 777 nm line and CCD at Spectrum (2014/15)
⇒ Matrix independent for Cu-, Mg- and Al-matrix

Quantification of Oxygen



Use of 777 nm line and CCD at Spectruma (2014/15)
⇒ Matrix independent for Cu-, Mg- and Al-matrix

Spectruma

Digikröm monochromator (2015)

- Installation of new grating (1200 g/mm), second order filter (500 nm) and NIR-PMT (R6357 Hamamatsu)
⇒ Resolution@777 nm > 20 pm

GDA750 polychromator (2016)

- Installation of new optics with HR CCD
- Installation of NIR-PMT @777 nm

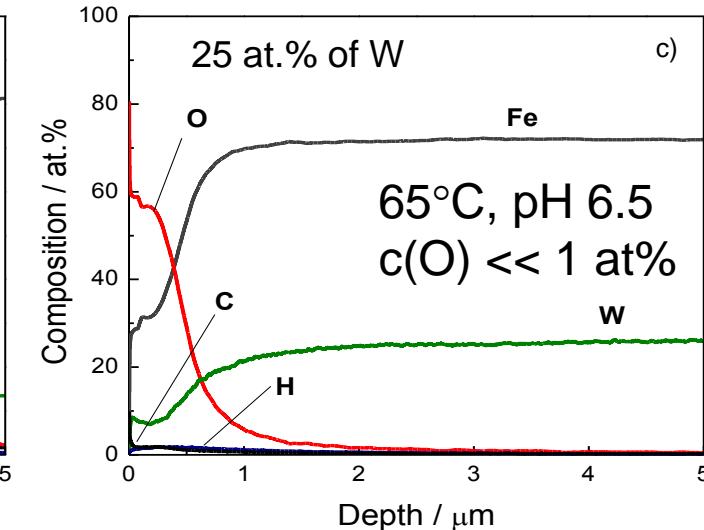
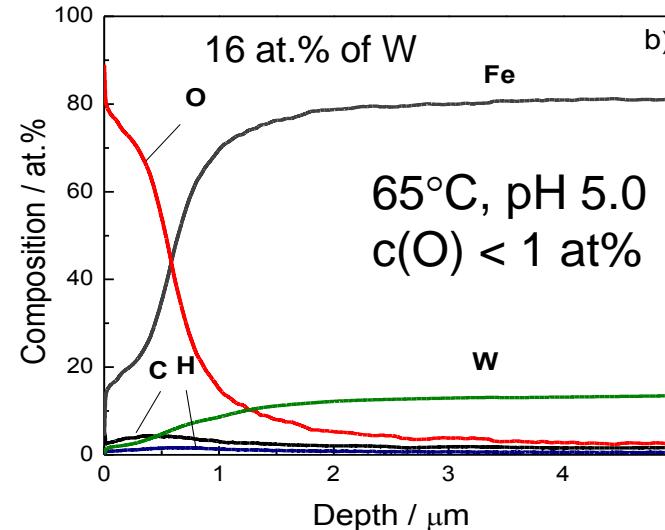
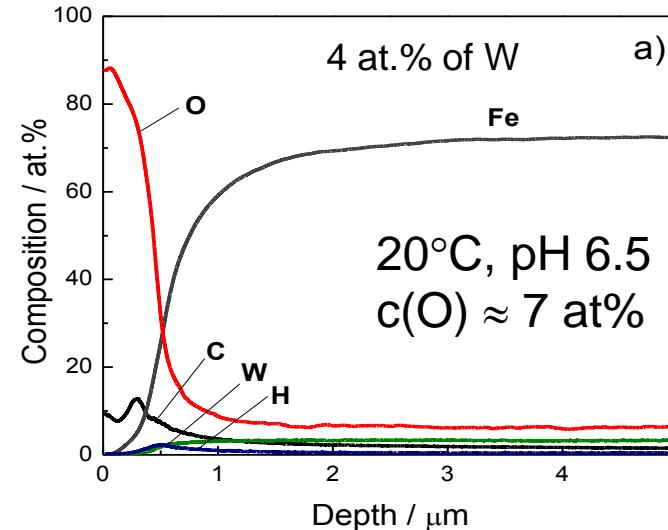
⇒ Dark current of NIR-PMTs depends on T
⇒ use of CCD for application

Analysis of Electrochemically Deposited FeW

A. Mulone, A. Nicolenco, V. Hoffmann et al., *Electrochimica Acta* 261, 2018, 167

≈ 500 nm FeO_x (C,H,W) and FeW layer for HT applications

1.0 glycolic acid, 0.3 citric acid, 0.1 $\text{Fe}_2(\text{SO}_4)_3$ and 0.3 Na_2WO_4 , NaOH and H_2SO_4 for pH regulation



c(W) and c(O) in FeW_x depend on T and pH at deposition and are essential for HT behaviour.

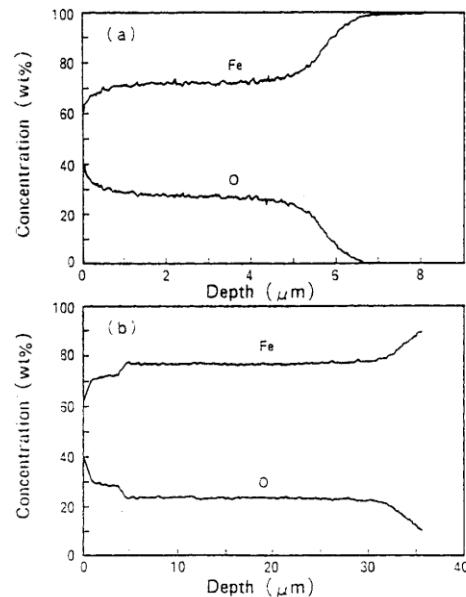
1. Calibration correct for Fe?
2. Application of PMT possible for thin layers or low c(O)?

FeO-layers from Literature

S. Suzuki, K. Suzuki, K. Mizuno from Nippon Steel, Japan

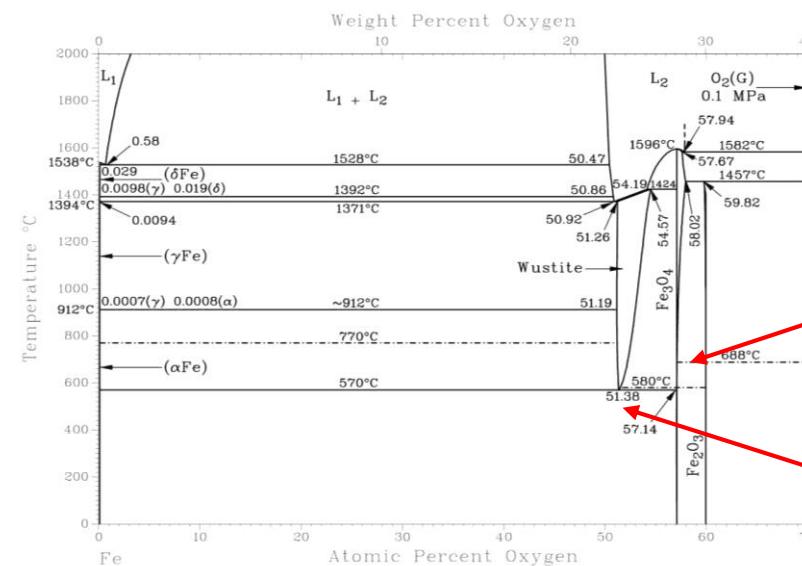
Quantitative GDOS Analysis of Oxide Films on Steels, Surf. Interface Anal., 22 (1994) 134-138.

- Matrix specific calibration of 130 nm O-line with Fe_3O_4 -layer (background/zero not reported)
- Fe_3O_4 - and FeO-layers on construction steel in air at 600°C and 800°C (time not reported)
- Comparison of EY with HIP of Fe+MnO and Fe+SiO₂, (c not reported) $\Rightarrow \pm 5\%$ same EY



Magnetit
6 μm Fe_3O_4 -layer used
for 1 point calibration

Wüstite
„thick“ FeO-layer



$\text{Fe}_3\text{O}_4/\text{Fe}_2\text{O}_3$ -layer
 $c(\text{O})=57-60$ at%
(27-30 m%)

FeO-layer
 $c(\text{O})=51.38$ at%
(23.19 m%)

Figure 5. Depth profiles of Sample A, oxidized (a) at 873 K and (b) at 1073 K which were corrected based on the present method.

(C) 1996 ASM International

Preparation of FeO-layers at IFW



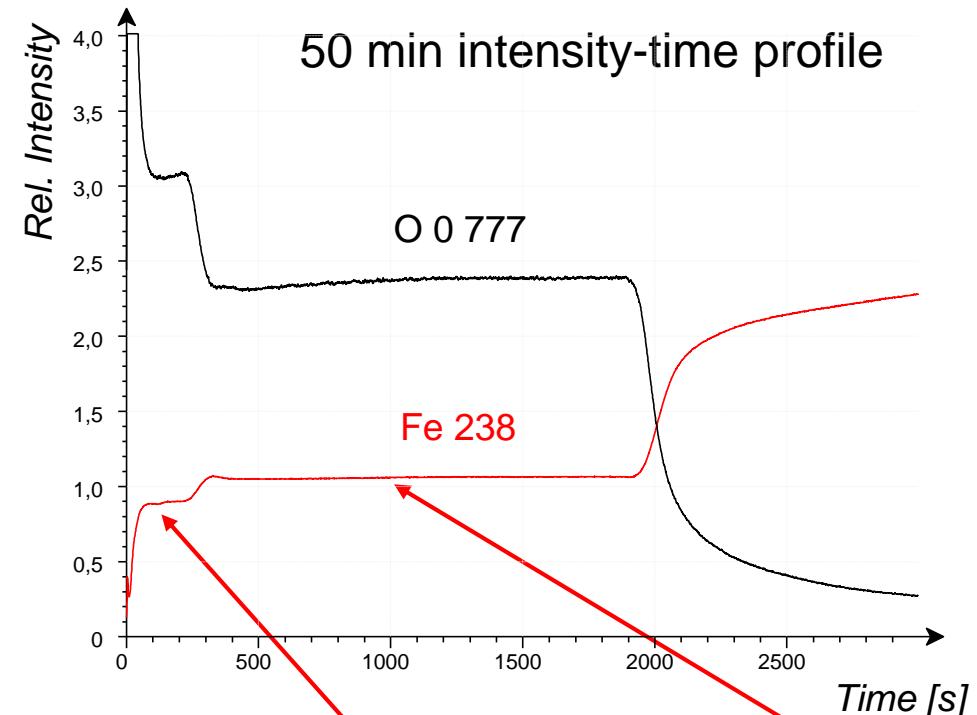
1200°C, 1 h, air

800°C, 10 min, air

800°C, 10 min, 200 mbar O₂
6.5 µm Fe₃O₄ + 43 µm FeO

Construction steel
> 99 m% Fe

phases (XRD)	Fe ₃ O ₄ /Fe ₂ O ₃	FeO
d(Profilometer), µm	6.5 µm	43 µm
c(O, RBS), m%	60 at%	52 at%

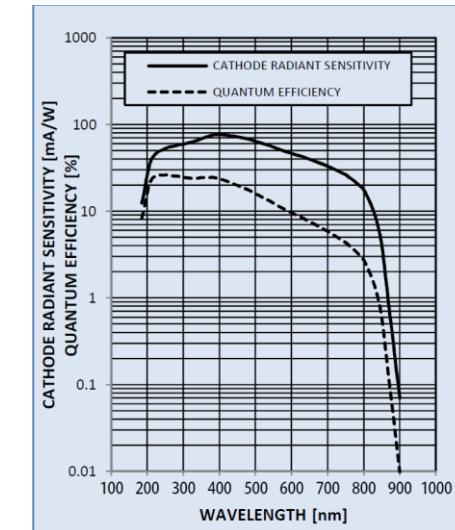
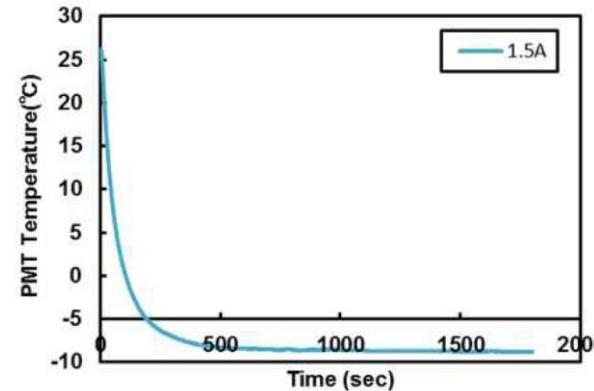


PMT H14768 (Hamamatsu) with Peltier cooling

Installation at monochromator Digikröm (Spectral Products)

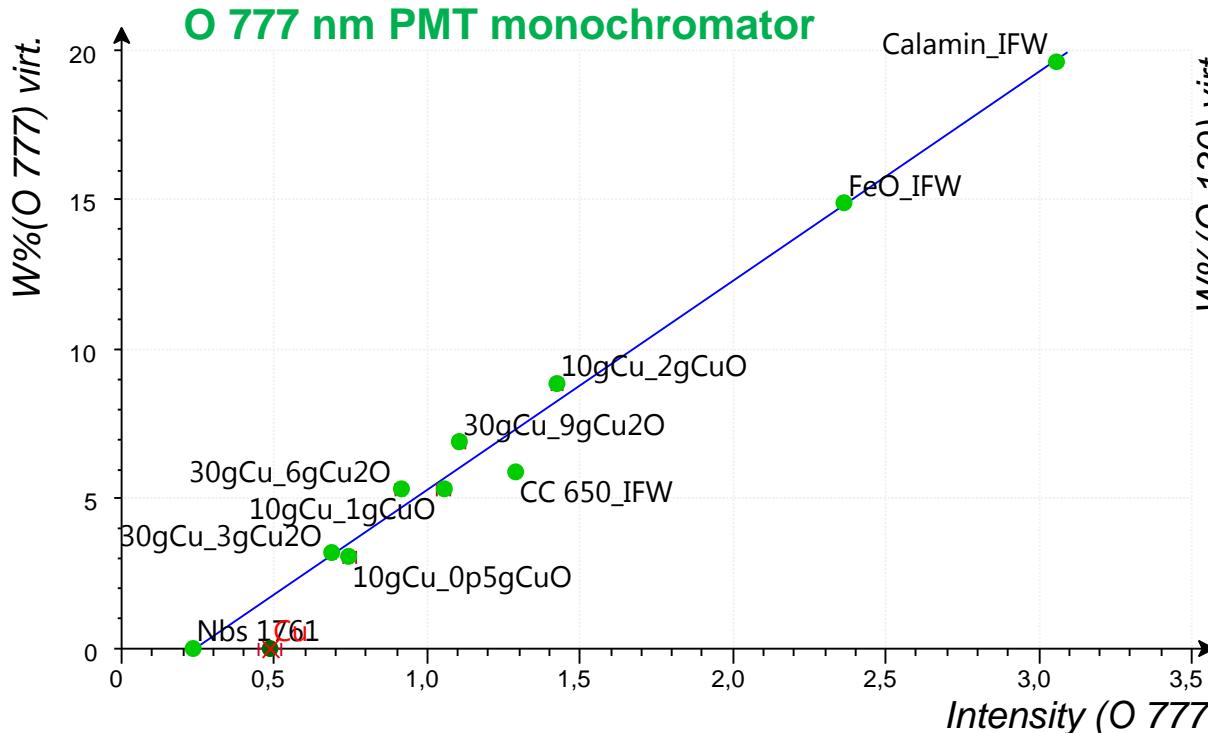


Size: (W) x (D) x (H)
64 x 64 x 134 mm



Figures from H14768_Technical Information_181210.pdf (Hamamatsu)

Comparison of O-Calibration at 777 nm and 130 nm with PMT

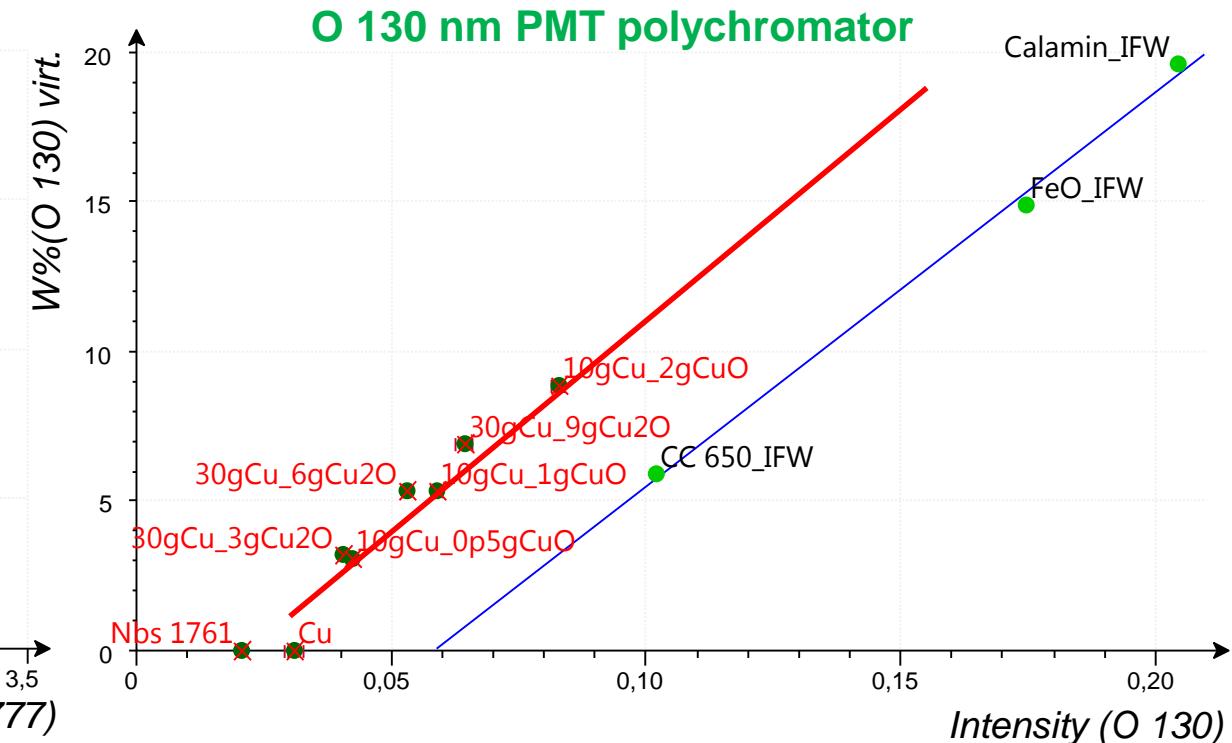


900 V PMT voltage

⇒ 777 nm line is more sensitive, EY(O) similar in Cu, Al, Mg, Fe and CC650

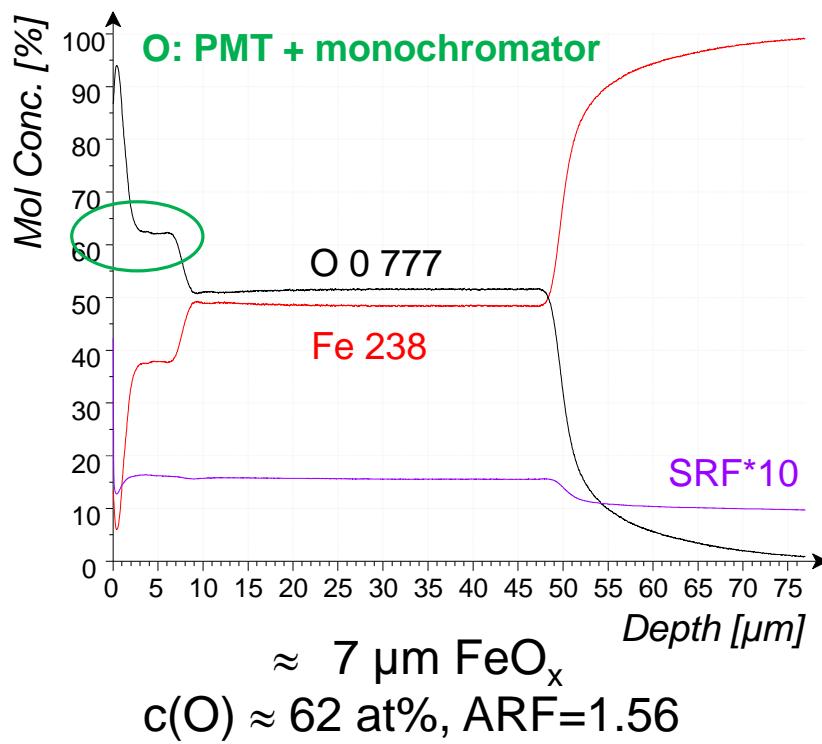
⇒ Less scattering at 130 nm, but clear dependence of EY on matrix

⇒ Background depends on matrix at both lines

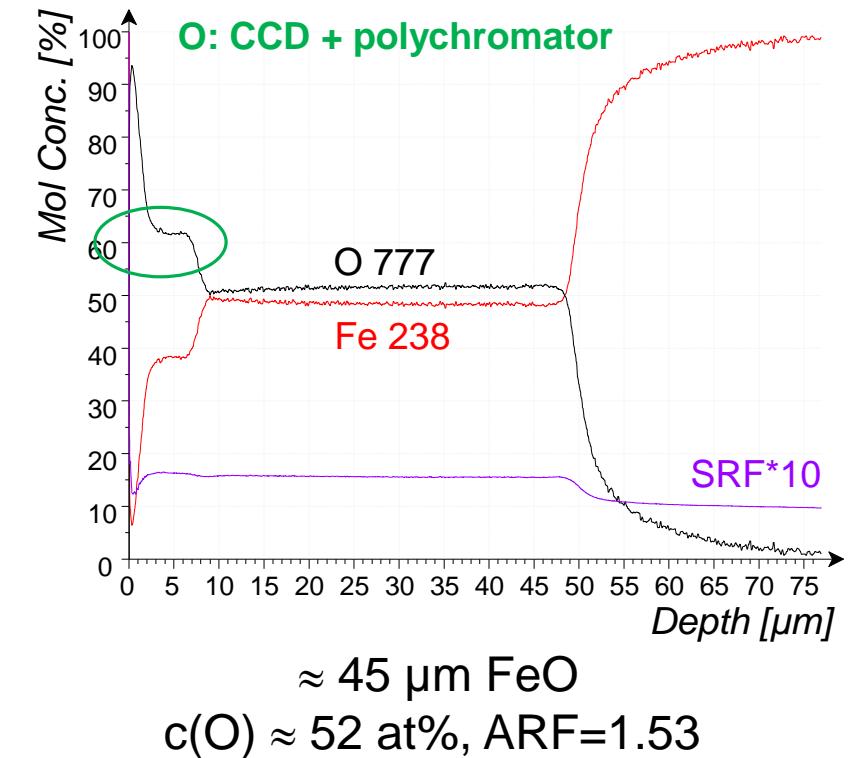


HS3 ≈ 900 V PMT voltage

Compositional Depth Profile of FeO-layers from IFW



$$\begin{aligned}\rho(\text{O, Fe}_3\text{O}_4) &= 2.75 \text{ g/cm}^3 \\ &< \\ \text{used } \rho(\text{O}) &= 2.9 \text{ g/cm}^3 \\ &< \\ \rho(\text{O, FeO}) &= 3.14 \text{ g/cm}^3\end{aligned}$$



→ Cooled PMT H14768 successfully coupled to monochromator

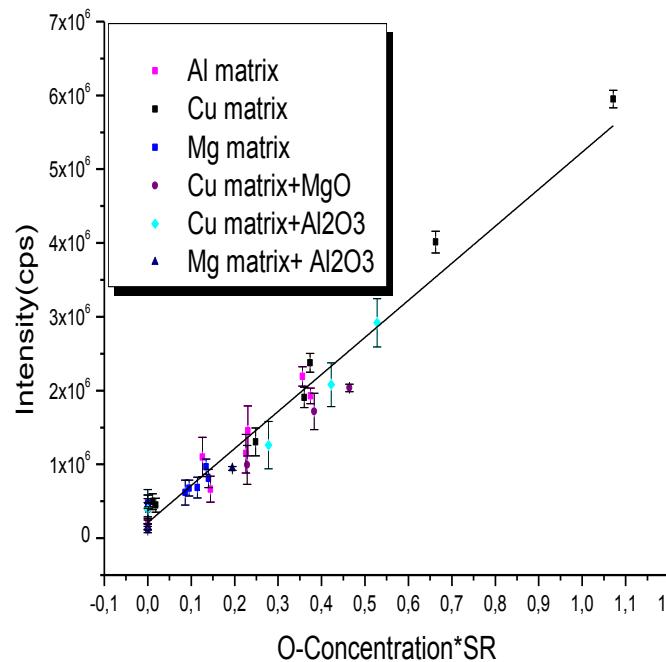
Possible improvements: reduction of O background and scattering, better Fe calibration

Calibration of Oxygen **in GD-MS**

Main investigations with Element GD at Evonik (VG9000 for comparison) using a set of certified Al, Cu and Zn samples. (*Evonik, BAM, IFW @ EMRP project*)

- At constant flow and current (400 sccm and 50 mA) StdRSF method and absolute method are matrix-independent.
 - Deviations < 15% from reference value in the middle mass range.
 - StdRSFs(Cu, Al) depend on samples or concentration?
 - The stability of discharge (especially, if Oxygen is present at higher concentrations) is much better at **matrix-specific conditions**, which is needed for good intensity and sputtering rate measurements. Sensitivity must be checked \Rightarrow deviations < 15%
- \Rightarrow Calibration of Oxygen at **matrix-specific conditions** in absolute mode

Multi-matrix calibration for Oxygen at Element GD and VG9000



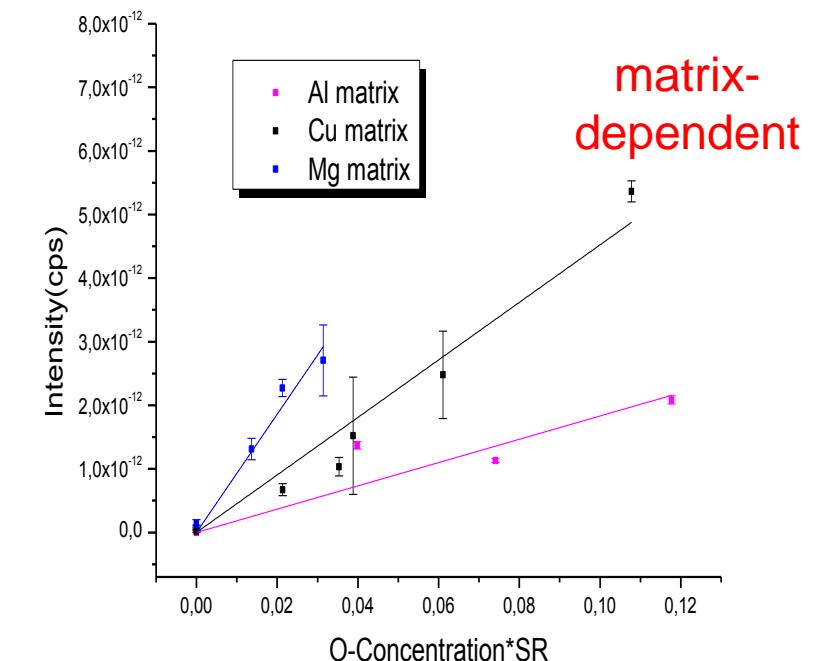
ELEMENT GD
Matrix-specific conditions
Al: 320 sccm, 80 mA
Cu: 400 sccm, 50 mA
Mg: 320 sccm, 80 mA

Deviations for c(O) < 15%
using the absolute model and
matrix specific conditions

Sensitivity of O is independent
of chemical form of the oxide
and matrix

C. Gonzalez-Gago et al., Investigations of matrix independent calibration approaches in fast flow glow discharge mass spectrometry, J. Anal. At. Spectrom., 2019, 34, 1109-1125
(Evonik, BAM, IFW @ EMRP project)

VG9000
Matrix-specific conditions
Al: 5 mA, 1.25 kV
Cu: 5 mA, 1 kV
Mg: 3 mA, 1 kV



Latest Publication about Quantification in GD-MS

C. Gonzalez-Gago et al., *Investigations of matrix independent calibration approaches in fast flow glow discharge mass spectrometry*, J. Anal. At. Spectrom., 2019, 34, 1109-1125 (Evonik, BAM, IFW @ EMRP project)

Element GD from Thermo, flat samples, systematic study of quantification using Cu, Al and Zn standards

G. Paudel, M. Kasik, M. Di Sabatino, *Investigation of the intensity dependence of glow discharge mass spectrometry quantification on the discharge parameters*, J. Anal. At. Spectrom., 2019, 34, 1829-1837

Astrum instrument from Nu, pin samples, systematic study of discharge parameter changes using a Ta sample

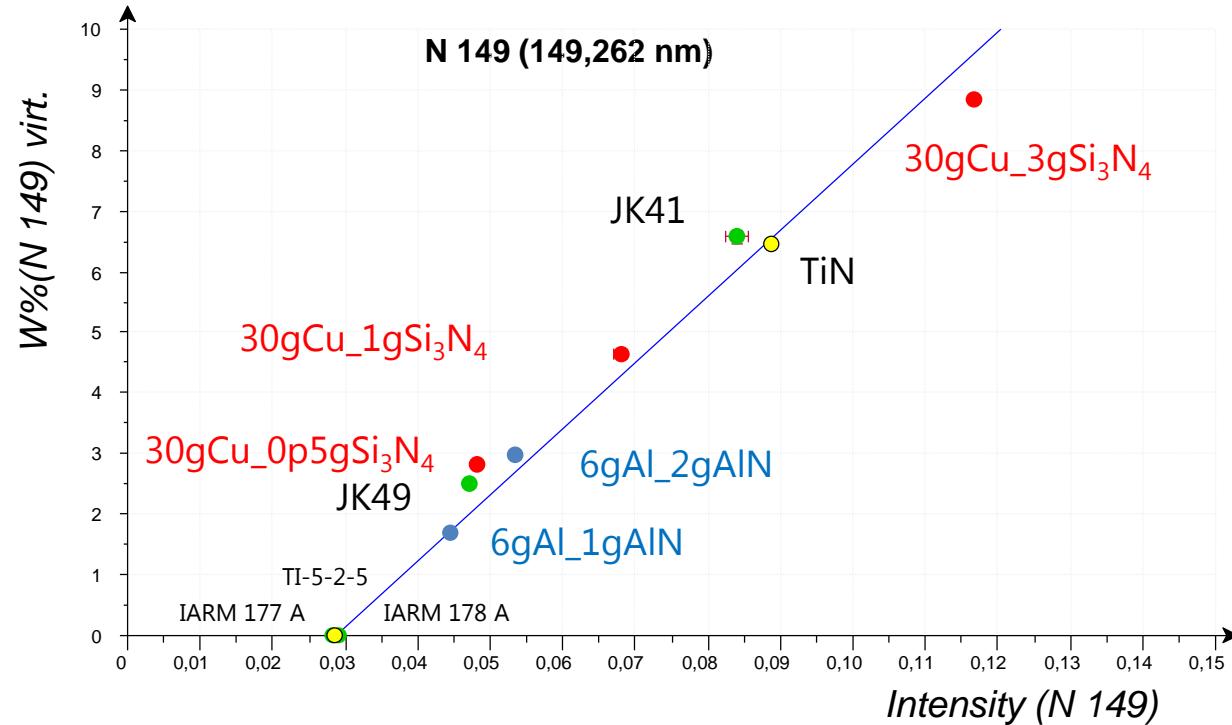
⇒ **RSD ≤ 5%**, discharge parameters must be considered at quantification

⇒ Continuation of investigations, instrumental and software improvements are needed, including Astrum from Nu and considering

T. Saka, M. Yamaguchi, K. Ito, *Concentration Dependence of Relative Sensitivity Factors in Glow Discharge Mass Spectrometry*, Anal. Sci., 2001, 17, i841-i844

VG9000 from former VG, flat and pin samples ⇒ Cu and Al matrix show deviations using the StdRSF concept assuming non-linearities of RSFs.

Results for Nitrogen

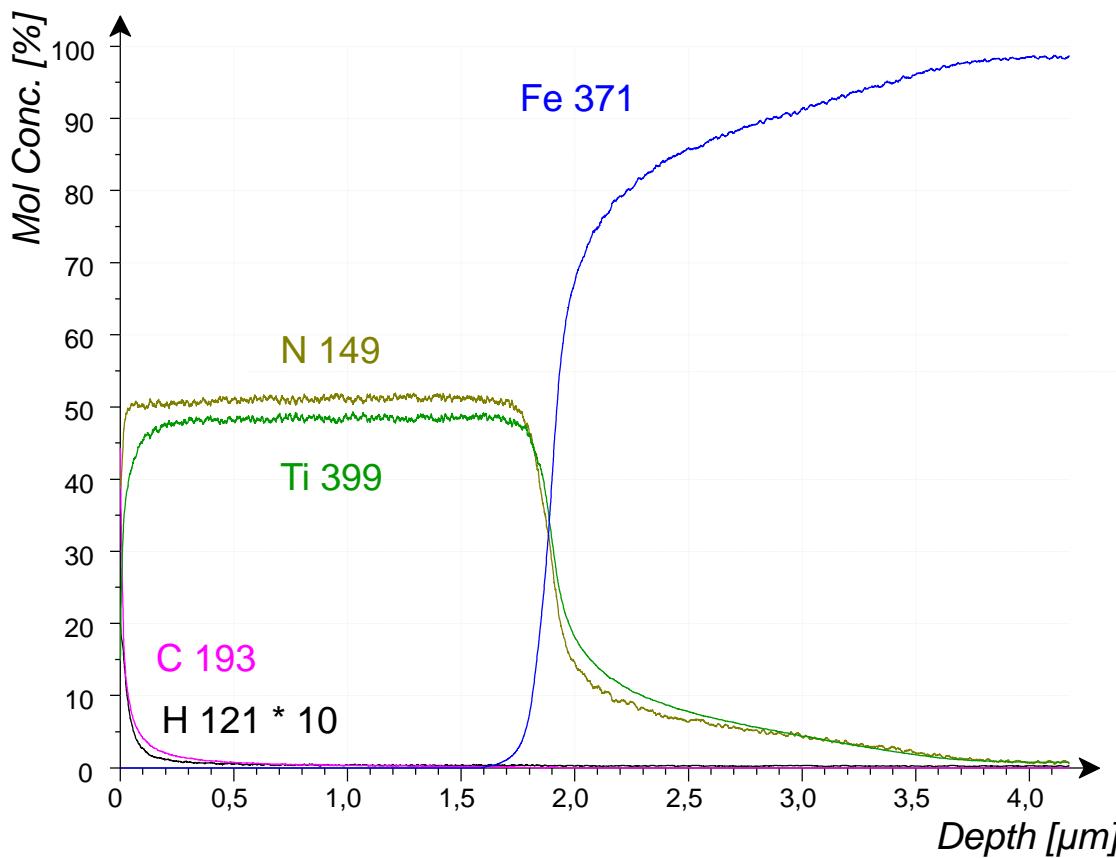


GD-OES Calibration at 149 nm

- 4 mm DC source, 700 V, 20 mA
- Validated c(N) for Cu + Si₃N₄
- EY(N) of sintered Cu + Si₃N₄ agrees with
 - EY(N) in sintered Al + Al₂O₃
 - EY(N) in TiN and in certified
 - JK41 (c(N)=6.9 m% - out of stock)
 - JK49 (c(N)=1.89 m%)

⇒ No dependence on matrix (Al, Cu, Ti, Fe) and chemical form of the nitride was observed.

Quantification of $\approx 2 \mu\text{m}$ thick TiN-Layer



4 μm sputtered in 5 min

$\Delta c(\text{N, Ti}) \approx 5\%$ after 100 nm
(C_xH_y contamination)

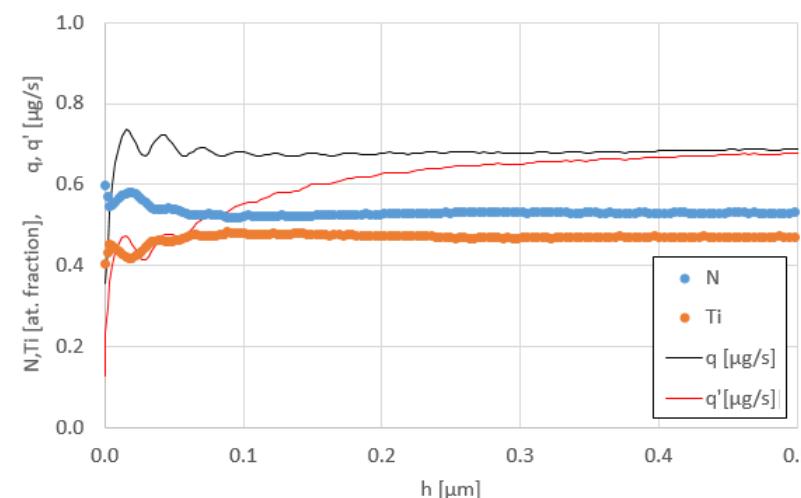
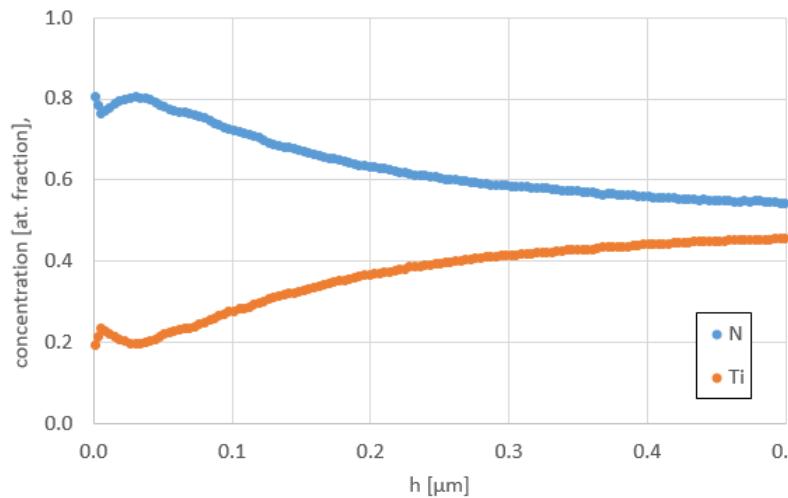
$\rho(\text{N, TiN}) = 11.3 \text{ g/cm}^3$ was used

Hard-core model

$$\frac{1}{\rho} = \sum \left(\frac{c_i}{\rho_i} \right) \quad z = \sum \left(\frac{q}{\rho \cdot A} \cdot \Delta t \right)$$

Effect of other light Elements on the Analysis of TiN-Layers

- Arne Bengtson, H-effect
- Vasile-Dan Hodoroaba, Volker Hoffmann, Edward Steers, H-effect
- Richard Payling, Max Aeberhard, Daniel Delfosse, 2001, J. Anal. At. Spectrom., 2001, 16, 50-55, Improved quantitative analysis of hard coatings by rf-GD-OES
- ...



Influence of H, C on Ti, N until 500 nm \Rightarrow Interelement effects must be corrected

Zdenek Weiss, Petr Vlcak, J. Anal. At. Spectrom., 2017, 32, 2476–2484

Analysis of shallow depth profiles of titanium nitride and N-implanted titanium by GD-OES:
the ‘hydrogen effect’ after the discharge startup and a correction thereof

Summary

1. Hydrogen

- Absolute calibration in GD-OES (< 1 m%) and GD-MS (< 1.5 m%)
- No matrix effect until now (CuTi, CuZr)

7 sets (3 samples) with Cu+ TiH_2 prepared with $c(\text{H}) = 0.13 - 0.36 \text{ m\%}$
0.0331 and 0.0657 m% positively tested

2. Oxygen

- EY(O 130 nm) is matrix **dependent**
- EY(O 777 nm) in GD-OES **and** sensitivity(O) in fast flow GD-MS
are **independent of matrix** (Al, Cu, Mg) and **chemical form of oxide**
- Cooled PMT H14768 successfully coupled to monochromator

7 sets with Cu+ Cu_2O prepared with $c(\text{O}) = 1.1 - 2.6 \text{ m\%}$

3. Nitrogen

- EY(N 149 nm) is independent of matrix (Cu, Al, Ti, Fe) and chemical form of nitride
- $c(\text{N})$ of Cu+ Si_3N_4 is validated. $c(\text{N}) = 0.9 - 3.5 \text{ m\%}$

Much progress at light element analysis by GDS

Acknowledgement

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Wolfgang Löser, Andrea Voss

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Annett Gebert, Aliona Nicolenco

SIB09-Primary Standards for Challenging Elements

BAM

Silke Richter, Jens Pfeifer

Heinrich Kipphardt



AQura / Evonik

Cristina Gonzalez-Gago, Petr Smid

Thomas Hofmann, Cornel Venzago



HZDR René Heller

Thank you for your attention!